

Equations for the Exam on Unit T

$dU = mc dT$	$PV = Nk_B T$	$U = \frac{f}{2} Nk_B T$	$\Delta U = Q + W$
$K_{avg} = \frac{1}{2} m [v^2]_{avg} = \frac{3}{2} k_B T$	$v_{rms} = \sqrt{[v^2]_{avg}}$	$W = -\int P dV$	$W = -Nk_B T \ln \frac{V_f}{V_i}$
adiabatic:	$TV^{g-1} = \text{constant}$	$PV^g = \text{constant}$	
$W(N, U) = \frac{(q + 3N - 1)!}{q!(3N - 1)!}$	$q = U/e$	$W_{AB} = W_A W_B$	$S = k_B \ln W$
$S_{AB} = S_A + S_B$	$\frac{1}{T} = \frac{dS}{dU}$	$\Delta S = mc \ln \frac{T_f}{T_i}; \frac{V_f}{V_i}$	$dS = \frac{dQ}{T}$
$\text{Pr}(E) = \frac{1}{Z} e^{-E/k_B T}$	$Z = \sum_{\text{all states}} e^{-E_i/k_B T}$	$E_{avg} = \sum E_n \left(\frac{e^{-E_n/k_B T}}{Z} \right)$	$Q = mL$
$v_p = \sqrt{\frac{2k_B T}{m}}$	$D(v) = \frac{4}{\sqrt{\rho}} \left(\frac{v}{v_p} \right)^2 e^{-(v/v_p)^2}$	$\text{Pr}(v_1 < v < v_2) = \int_{v_1}^{v_2} D(v) \frac{dv}{v_p}$	
$e = \frac{ W }{ Q_H } \leq \frac{T_H - T_C}{T_H}$	$\text{COP} = \frac{ Q_C }{ W } \leq \frac{T_C}{T_H - T_C}$		

Physical Constants and Data

$k_B = 1.38 \times 10^{-23} \text{ J/K} = 8.62 \times 10^{-5} \text{ eV/K}$	$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J}$
$N_A = 6.02 \times 10^{23} \text{ molecules/mole}$	$m_{\text{proton}} \gg m_{\text{neutron}} \gg 1.7 \times 10^{-27} \text{ kg}$
$m_e = 9.11 \times 10^{-31} \text{ kg}$	$1 \text{ atm} = 101.3 \text{ kPa}$

Avogadro's number of nucleons (protons and/or neutrons) has a mass of about 1 g

$g = 5/3$ (for monatomic gas) $g = 7/5$ (for diatomic gas)

a monatomic gas has 3 degrees of freedom; a diatomic gas has 5 degrees of freedom

specific heat of water = 4186 J/(kg·K) latent heat of melting ice = 333 kJ/kg

Propagation of Uncertainties

In general: $f(a, b, \dots)$

$$U[f] = \sqrt{\left(\frac{\partial f}{\partial a} U[a] \right)^2 + \left(\frac{\partial f}{\partial b} U[b] \right)^2 + \dots}$$

If $f = \frac{ab \dots}{cd \dots}$

$$U[f] = |f| \sqrt{\left(\frac{U[a]}{a} \right)^2 + \left(\frac{U[b]}{b} \right)^2 + \left(\frac{U[c]}{c} \right)^2 + \dots}$$

If $f = a + b + \dots - c - d - \dots$

$$U[f] = \sqrt{(U[a])^2 + (U[b])^2 + (U[c])^2 + \dots}$$